The infrared spectrum of H_2S from 1 to 5 μ m

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This paper is dedicated to Dr. Gerhard Herzberg on the occasion of his 90th birthday

The absorption spectra of $\rm H_2S$ from 2000 to 11 147 cm⁻¹ have been obtained with spectral resolutions of 0.006, 0.012, and 0.021 cm⁻¹ using the Fourier transform spectrometer at Kitt Peak National Observatory. The transitions of 21 bands have been assigned for the first time and 9 others reanalyzed so that accurate energy levels, band origins, and rotational parameters could be determined. The analysis of these data revealed some remarkable features in the energy spectrum, e.g., fourfold clustering of rotational levels belonging to the symmetric and asymmetric components of local mode manifolds at a high degree of stretching excitation. This paper reports fitted vibrational parameters and predicted band origins of $\rm H_2^{32}S$ up to 12 735 cm⁻¹. It also presents the degenerate rotational constants and upper state energies of (301)–(202) and (311)–(212) at 1 μ m as illustrations of clustering in the local mode limit.

Les spectres d'absorption de H₂S entre 2000 et 11 147 cm⁻¹ ont été obtenus avec des résolutions spectrales de 0,006, 0,012 et 0,021 cm⁻¹, en utilisant le spectromètre à transformée de Fourier du Kitt Peak National Observatory. Les transitions de 21 bandes ont été identifiées pour la première fois, et 9 autres ont été réanalysées de façon à pouvoir déterminer avec précision les niveaux d'énergie, les origines de bandes et les paramètres rotationnels. L'analyse de ces données a révélé quelques particularités remarquables du spectre d'énergie, entre autres le groupement quadruple de niveaux rotationnels appartenant aux composantes symétriques et asymétriques d'ensembles de modes locaux à un haut degré d'excitation d'allongement. On donne dans cet article des valeurs ajustées des paramètres vibrationnels, ainsi que des prédictions pour les origines de bandes de H₂, ³²S jusqu'à 12 735 cm⁻¹. On présente aussi les constantes rotationnelles dégénérées et les énergies des états supérieurs de (301)–(202) et (311)–(212) à 1 µm, comme illustration du groupement à la limite des modes locaux.

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1. Introduction

The detailed knowledge of hydrogen sulfide absorption spectra has application for terrestrial atmospheric pollutant measurements and for the investigation of chemistry in the atmospheres of Venus and the outer planets. From a theoretical viewpoint, hydrogen sulfide is an interesting example of a light asymmetric rotor for which the internal nuclear motion can be strongly perturbed by intramolecular interactions arising from vibrational or rotational excitation. For this reason, the vibrational-rotational energy spectrum of H₂S has been modeled in numerous papers using new theoretical approaches to demonstrate the effects of the local mode vibrations or bending-rotation coupling [1-3]. The infrared spectrum of H₂S has been the subject of several high-resolution studies concerning the ground vibrational state [4, 5], the first excited level (010) at 8.3 µm [6, 7], the first triad of interacting states $\{(020)-(100)-(001)\}\$ at 4 μ m [8], two levels $\{(110)-(011)\}\$ from the second triad at 2.7 μ m [9], and the {(101)–(200)} [10] and {(111)-(210)} [11] states belonging to the first and second hexades at 2 µm and 1.6 µm, respectively. However, up to now, the knowledge of hydrogen sulfide absorption and its energy spectrum has been incomplete, especially in the case of weak overtone stretching and bending modes. The lack of experimental data has limited both the theoretical analysis and the prediction of the near-infrared and the visible regions.

The present study reports the vibrational assignment of H₂³²S over a wide spectral interval from 2000 to 11 147 cm⁻¹. Figure 1

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shows a composite spectrum obtained from two different optical densities. In all, transitions of a total of 30 vibrational bands have been observed. Of these, 21 bands, including (311) at 11 008 cm $^{-1}$, have been identified at high-resolution for the first time. In this report, we present the vibrational energy levels analysis along with the rotational assignments at 1 μm of pairs of parallel and perpendicular bands that become rotationally degenerate in the local mode limit.

2. Experimental details

Laboratory spectra of H₂S were recorded at 0.006, 0.012, and $0.020 \, \mathrm{cm}^{-1}$ resolution with the Fourier spectrometer located at the McMath telescope facility at Kitt Peak National Observatory/National Solar Observatory. Data were obtained using three different beamsplitters (KCl, CaF₂, and quartz) in conjunction with As-doped silicon, InSb, and photo-diode detectors in five different band pass intervals: $1000 - 2600 \text{ cm}^{-1}$, $1800 - 5500 \text{ cm}^{-1}$, $3600 - 9000 \text{ cm}^{-1}$, $3600 - 9000 \text{ cm}^{-1}$ $13~000~cm^{-1}$, and $8600-16~000~cm^{-1}$. The optical sources were a globar at longer wavelengths and a quartz projection lamp in the near-infrared and visible regions. Optical path lengths were changed from 1.5 m to a maximum of 433 m by using three different stainless steel absorption cells. Sample pressures were varied from 1.49 to 30 Torr (1 Torr = 133.3 Pa) at room temperature. For some scans, a second absorption cell containing CO was placed in series with the H₂S cell to establish the frequency calibration in the near-infrared using the 2-0 positions reported by Pollock et al. [12]. Each spectrum was usually integrated 70-80 min to produce signal-to-noise ratios ranging from

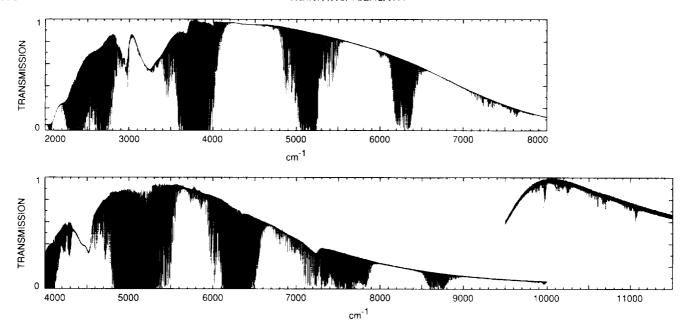


FIG. 1. The observed spectra of H_2S recorded at two optical densities. Top: 28.5 m with 9.99 Torr at 289 K. Bottom: 433 m with 29.9 Torr at 298 K. The upper panel shows the portions from the 1800-5500 and 3600-9000 cm $^{-1}$ band passes, which are joined together at 4000 cm $^{-1}$. The lower frame is taken from the 3600-13~000 (left) and 8600-16~000 (right) cm $^{-1}$ band passes. Spectral impurities arise from H_2O at 3700, 5300, 7200, 8890, and 1060 cm $^{-1}$; CO_2 at 2350 and 3714 cm $^{-1}$; SO_2 at 2100, C_2H_6 at 2980, and HCN at 3300 cm $^{-1}$.

TABLE 1. Summation of assigned levels for 27 H₂S bands in the near-infrared*

System	Region (µm)	Band	$J_{ m max}$	$K_a(max)$	Number
Second triad	2.7	030	8	6	41
		110	10	7	94
		011	10	7	100
First hexade	2.0	040	9	7	54
		120	10	7	73
		021	11	9	115
		200	11	10	124
		101	11	10	128
		002	11	6	68
Second hexade	1.6	050	12	5	65
		130	8	7	63
		031	8	5 7 7	65
		210	8	7	67
		111	8	7	60
First decade	1.3	121	7	6	65
		102	8	7	75
		201	8 8	8 8	74
		300	8	8	75
		003	8	8	76
Second decade	1.1	112	9	7	74
		211	9	8	86
First or second pentadecades	1	301	8	8	74
1		202	8	7	47
		400	9		58
		103	9	8	67
		311	8	8 8 8 7	63
		212	7	7	35

^{*}SYSTEM indicates the total number of interacting states: triad (3), hexade (6), decade (10), and pentadecade (15). BAND is the quanta of $\nu_1\nu_2\nu_3$ (110, 011, 101, 002, 210, and 111 have been previously studied at high resolution). NUMBER shows the total number of upper state levels assigned to maximum J and K_a .

BYKOV ET AL. 991

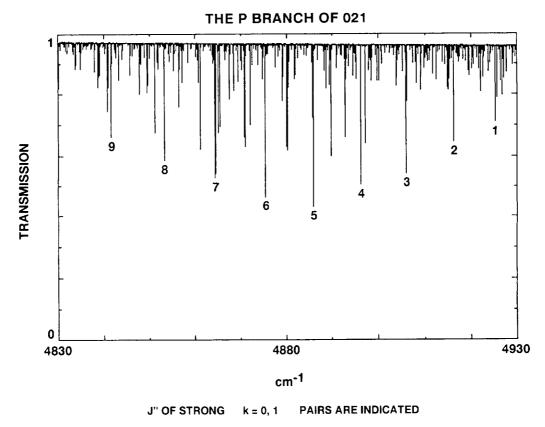


Fig. 2. An apodized Kitt Peak FTS spectrum of H_2S in the *P*-branch region of (021). The J'' assignments of the $K_a = 0.1$ lines are indicated. The optical path is 28.5 m, and the sample pressure is 9.99 Torr at 289 K.

1000:1 at 2.5 μm to 400:1 at 8 μm to 50:1 at 1 μm . Figure 1 gives a composite spectra obtained with the last four band passes: the top was recorded with a 28 m optical path and 9.99 Torr of H_2S at 289 K; the two band passes are joined together at $4000~cm^{-1}$. The bottom trace was obtained with a 433 m path and 29.9 Torr at 298 K.

The line centers were determined either by doing first and second derivatives of the apodized spectra or by least-squares fitting of the Voigt contour with the unapodized data. The precision and accuracy of a line center varied according to the region, gas pressure, and degree of blending with other features. Between 2000 and 5500 cm⁻¹, where the resolution was 0.012 cm⁻¹, the precision is 0.0002 cm⁻¹, and the accuracy is 0.0004 cm⁻¹ or better for strong, isolated lines. For weak lines and all lines above 6600 cm⁻¹, these values are worse by a factor of two because of the reduced signal-to-noise ratio. However, at 11 000 cm⁻¹ where the resolution was 0.021 cm⁻¹, the precision at best is 0.0010 cm⁻¹ because the signal-to-noise ratio was poor. In addition, higher sample pressures of 10–30 Torr were required to observe these very weak bands so that line centers were affected by pressure shifts. The absolute line positions were further degraded because no suitable calibration lines were available near 1 µm; for the interim, these data were calibrated using H₂S lines at 8800 cm⁻¹ recorded using the 3600 to 13 000 cm⁻¹ band pass with the same gas sample. Overall, the accuracy of the near-infrared positions between 8000 and 11 000 cm⁻¹ is estimated to be 0.005 cm⁻¹ or better.

3. Line assignments and rotational energy levels

The line assignments were made using the combination differences and estimated line frequencies and strengths as described in ref. 13. The line assignment process was followed by continual fitting of the rotational constants to obtain better predicted line positions and relative strengths. Such a procedure often permitted the identification of weak lines that could not be assigned by usual combination difference methods. Table 1 summarizes the upper vibrational states of 27 bands belonging to different interacting band systems (triad, hexade, decade, and pentadecade) that have been observed so far in the near-infrared portion of the Kitt Peak spectra. The extent of identification is given by indicating the largest values of the J and K_a quantum numbers $(J_{max} \text{ and } K_a(max))$ obtained up to now and the total number of upper state levels assigned. Figure 2 shows the P branch of (021) recorded at 0.012 cm⁻¹ resolution with a 28 m optical path and 9.99 Torr of H₂S at 289 K. Table 2 lists the observed band centers for 31 energy states; the value of the lowest fundamental is from ref. 7. All but five band origins were obtained using the observed assignment of P(1,1,1) or P(1,0,1)to the 0,0,0 levels. Table 2 also lists the experimental line positions and corresponding intensities (in cm⁻² atm⁻¹ at room temperature; 1 atm = 101.3 kPa) of the P1 lines. For the (040), (050), (202), (212), and (400) states, the band centers were determined by fitting available assignments because the P1 lines were too weak to be observed. The accuracy of these fitted values ranges from 0.002 to 0.005 cm⁻¹.

TABLE 2. Observed * and calculated vibrational upper state energy levels of $\mathrm{H_2^{32}S}$

992						CAN. J. PH	13. VOL. 72, 1994	
	Senekowitsch [15] ab initio	1 190.4	2 372.0 2 620.4 2 631.0	3 543.5 3 794.6 3 799.8	4 703.7 4 960.1 4 960.0 5 154.2 5 155.5 5 251.2	5 851.6 6 115.6 6 110.2 6 307.7 6 403.0	6 986.4 7 249.4 7 259.8 7 451.5 7 546.9 7 589.4 7 789.4	8 307.0 8 376.5 8 391.7 8 585.3 8 681.5 8 723.0 8 723.1 8 906.9 8 917.4 9 212.7 9 490.6
	Kozin [3] MORBID	1 182.44	2 353.83 2 614.66 2 628.56	3 513.17 3 779.29 3 789.66	4 659.48 4 932.91 4 939.82 5 145.52 5 147.12 5 243.38	5 791.83 6 074.50 6 078.05 6 288.37 6 288.99 6 385.89	6 909.29 7 203.39 7 203.10 7 420.03 7 419.92 7 517.74 7 576.45 7 752.40 7 775.40	8 314.90 8 314.90 8 317.74 8 538.93 8 636.52 8 696.48 8 877.73 8 898.66 9 411.66 9 411.66
vels of H2 -S	$P1 Int (cm^{-2}/atm \times 10^{-5})$	145.	47.6 29.3 3.9	6.8 282. 980.	0.2 1.8 26.0 45.3 211.	<.1 0.6 0.9 9.4 119.	1.9 14.4 10.9 1.1 2.2	0.4 4.1
pper state energy le	$\frac{P1\nu}{(cm^{-1})}$	1 167.4851	2 338.8743 2 599.3179 2 614.7079	3 498.6999 3 764.0764 3 775.5226	4 646.586 4 917.6091 4 925.3381 5 129.8961 5 133.4742 5 228.0113	5 782.144 6 059.4922 6 063.8491 6 273.0561 6 275.4272	7 406.3460 7 561.2915 7 562.7987 7 737.1743	8 682.052 8 683.409
and calculated vibrational upper state energy levels of H ₂ -S	(obsd. – calc.) (cm $^{-1} \times 10^{-3}$)	14	14 53 24	-12 -4	- 22 - 16 - 28 - 36 - 46	-65 -23 -13 -47	17 - 11 - 23 - 22	72 40
TABLE 2. Observed and	$\frac{E_{\rm obs}}{({\rm cm}^{-1})}$	1 182.5742	2 353.9644 2 614.4080 2 628.4551	3 513.7900 3 779.1665 3 789.2688	†4 661.6770 4 932.6992 4 939.1044 5 144.9862 5 147.2205 5 243.1014	[‡] 5 797.235 6 074.5823 6 077.5954 6 288.1462 6 289.1735	7 420.0923 7 576.3816 7 576.5450 7 752.2644 7 779.3195	8 697.142 8 697.155
T	$rac{E_{ m calc}}{({ m cm}^{-1})}$	1 182.561	2 353.950 2 614.355 2 628.431	3 513.789 3 779.179 3 789.273	4 661.699 4 932.715 4 939.122 5 145.014 5 147.256 5 243.055	5 797.300 6 074.585 6 077.597 6 288.160 6 289.220 6 386.119	6 920.212 7 204.320 7 204.410 7 419.818 7 420.074 7 518.451 7 576.395 7 576.555 7 775.288	8 030.058 8 318.912 8 321.811 8 539.438 8 539.684 8 697.070 8 697.115 8 880.066 8 899.270 9 126.455 9 420.992 9 426.944
	Local $nm+/-v$	00+1	00 + 2 $10 + 0$ $10 - 0$	00+3 10+1 10-1	$00 + 4 \\ 10 + 2 \\ 10 - 2 \\ 20 + 0 \\ 20 - 0 \\ 11 + 0$	00+5 10+3 10-3 20+1 11+1	00 + 6 10 - 4 10 - 4 20 - 2 20 - 2 30 + 0 21 + 0 21 + 0	00 10 10 10 10 10 10 10 10 10 10 10 10 1
	$v_1 = v_2 = v_3$	0 1 0	0 2 0 1 0 0 0 0 1	0 3 0 1 1 0 0 1 1	0 0 7 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	0 2 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	0 - 0 - 2 - 0 - 0 0 0 0 0 0 0 0 0 0 0 0	0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0
								'

BYKOV ET AL. 993

TABLE 2. (concluded)

Senekowitsch [15] ab initio	9 708.0 9 805.6 9 847.8 9 848.0 9 929.1	
Kozin [3] MORBID	9 645.95 9 745.45 9 805.37 9 805.50 9 910.77 9 910.75 9 992.23 10 188.86	11 007:74
$PI Int (cm^{-2}/atm \times 10^{-5})$	0.2	0.3
P1v (cm ⁻¹)	9 897.277	10 994.938
(obsd. – calc.) (cm ⁻¹ ×10 ⁻³)	15 7 7 15	- 28 - 30
$E_{ m obs}$ $(m cm^{-1})$	[‡] 9 911.023 9 911.023 [‡] 10 188.301 10 194.448	[‡] 11 008.684 11 008.684
$E_{ m calc}$	9 647.447 9 749.111 9 806.479 9 806.482 9 911.008 9 996.658 10 008.533 10 188.299 10 194.434	10 292.738 10 742.184 10 742.793 10 846.547 10 904.297 11 008.713 11 101.6729 11 106.729 11 294.332 11 294.332 11 294.332 11 294.332 11 294.332 11 294.332 12 194.888 12 193.457 12 193.457 12 193.457 12 188.023 12 188.023 12 188.023 12 189.678 12 189.678
Local $nm+/-v$	20 + 4 + 4 11 + 4 + 4 30 - 2 30 - 2 40 + 0 40 - 0 40 - 0 21 - 2 21 - 2 31 + 0 31 - 0 31 - 0	20 20 20 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4 4
23	3030155150	4 -00-00-000-40-000-44-00
5	440000000	D ~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~
12	404-48804-	

*The levels and P1 positions are in units of cm⁻¹ while the observed – calculated are in units of 10⁻³ cm⁻¹. The observed line intensities are in units of 10⁻⁵ cm⁻² atm⁻¹ at room temperature; the v₂ values are from refs, 6 and 7. Some of these intensities may be unreliable indicators of band strengths because the values were obtained from only one or two optical densities and because these lines may be overlapped by other transitions. Labeling of the levels differs from that of refs. 3 and 15: our labeling is based on the mixing coefficients originating from the Darling–Dennison resonance interactions between vibrational states and theirs on the contribution of the basis function. The local mode labels follow the scheme of ref. 20.

‡Energy levels for the (040), (400), (050), states were obtained by fitting rotational levels available with uncertainties of 0.005 cm⁻¹. The (050) state was assigned during the revision of the manuscript and was therefore not included in the vibrational fit.

TABLE 3. Vibrational spectroscopic constants of H₂³²S*

Parameter	Value (cm ⁻¹)	Parameter	Value (cm ⁻¹)
ωι	2719.1770 (760)	$y_{122} \times 100$	-5.82 (190)
ω2	1212.840 (170)	y123	-1.1416(240)
ω3	2735.8241 (860)	$v_{222} \times 100$	-6.324(850)
X11	-24.2588(120)	$v_{233} \times 100$	8.87 (200)
X12	-17.0492(800)	J. 2	` '
X13	-94.9594(210)	$\Gamma_{ m DD}$	-23,274 98 (790)
X22	-5.3160(740)	γ2	0.475 7 (120)
X23	-21.3253(860)	,-	(,
X33	-24.4936(160)		

*Estimated uncertainties in the last digits (in parentheses) are one standard deviation. Number of levels, 30. Number of parameters. 15: rms, cm⁻¹, 0.037. Max. deviation, cm⁻¹, 0.072.

The intensities are given as an indication of relative band strengths. However, a number of caveats apply. The values are determined in normal isotopic abundance, rather than for a 100% H₂³²S sample. At best (for the 4 μm triad), the accuracies are \pm 5%. Except for the first triad, these intensities were retrieved from only one or two optical densities, rather than the usual five or six scans; therefore they may have a systematic offset of perhaps $\pm 20\%$ that will require correction after more spectra are recorded. Intensities of lines weaker than 10⁻⁵ cm⁻² atm⁻¹ are accurate to only $\pm 30\%$. The reported absorption intensity may contain contributions from other transitions and thus be too high. Finally, even if the measurement is good, caution should be exercised in extrapolating one intensity into a band strength. We note that Lechuga-Fossat et al. [8] required a seven-term dipole moment expansion to fit measured intensities of the first triad; the higher overtones are expected to have many resonances that greatly perturb intensities. Additional intensity measurements are planned, and their analyses will be included in a series of "polyad" articles. Nevertheless, the intensities listed in Table 2 correctly indicate a surprising result seen qualitatively in Fig. 1 and in refs. 8 and 11; in $\rm H_2S$, the combination bands are relatively strong compared to the fundamentals.

4. The vibrational energy levels of H₂³²S

The observed band origins have been used to determine the effective vibrational Hamiltonian constants and to calculate the highly excited vibrational levels. The effective vibrational Hamiltonian is the "spectroscopic Hamiltonian" of ref. 14 that includes the high anharmonic terms:

$$H = \sum_{ij} H_{ij} |i\rangle \langle j|$$

where

$$H_{ii} = \sum_{\lambda} \omega_{\lambda} \left(\upsilon_{\lambda}^{i} + \frac{1}{2} \right) + \sum_{\lambda, \mu \geq \lambda} x_{\lambda, \mu} \left(\upsilon_{\lambda}^{i} + \frac{1}{2} \right) \left(\upsilon_{\mu}^{i} + \frac{1}{2} \right) + \sum_{\lambda, \mu \geq \lambda, \nu \geq \mu} y_{\lambda \mu \nu} \left(\upsilon_{\lambda}^{i} + \frac{1}{2} \right) \left(\upsilon_{\mu}^{i} + \frac{1}{2} \right) \left(\upsilon_{\nu}^{i} + \frac{1}{2} \right) \cdot \dots$$

$$H_{ij} = \left\{ \Gamma_{DD} + \gamma_{1} \left(\upsilon_{1} + \frac{1}{2} \pm 2 \right) + \gamma_{2} \left(\upsilon_{2} + \frac{1}{2} \right) + \gamma_{3} \left(\upsilon_{3} + \frac{1}{2} \mp 2 \right) \right\} \left\{ \left(\upsilon_{1} + \frac{1}{2} \pm \frac{1}{2} \right) \left(\upsilon_{1} + \frac{1}{2} \pm \frac{3}{2} \right) \left(\upsilon_{3} + \frac{1}{2} \mp \frac{3}{2} \right) \left(\upsilon_{3} + \frac{1}{2} \mp \frac{1}{2} \right) \right\}^{1/2}$$

if

$$|i\rangle = |\mathbf{v}_1 \ \mathbf{v}_2 \ \mathbf{v}_3\rangle, \quad |j\rangle = |\mathbf{v}_1 \pm 2 \ \mathbf{v}_2 \ \mathbf{v}_3 \mp 2\rangle$$

Darling-Dennison resonance

The spectroscopic parameters determined are the harmonic frequencies ω_1 , ω_2 , and ω_3 , anharmonic x_{ij} , y_{ijk} constants, and the coupling Darling–Dennison resonance constants $\Gamma_{\rm DD}$ and γ_2 . The Fermi resonance was not included in the calculations because it was found to be insignificant. The fitted vibrational parameters of hydrogen sulfide are shown in Table 3 together with estimated uncertainties (one standard deviation). It may be seen that all the fitted constants are generally well determined. However, some of these $(\omega_2, x_{22}, y_{222})$ have correlation coefficients up to 0.9.

Table 2 shows the vibrational energy levels of H₂S up to 13 000 cm⁻¹; the nine columns give the normal and local mode assignments, the fitted and experimental values and the differences, the line positions and intensities of the observed P1

lines, and calculated values from refs. 3 and 15. The local modes are ascribed in the manner used by ref. 20. The maximum difference between observed and calculated vibrational energy levels is equal to 0.072 cm⁻¹, and the standard deviation is 0.038 cm⁻¹ for 31 observed band centers. Although the fit does not reproduce the band centers to their experimental accuracies, it has to be emphasized that this vibrational energy levels calculation is the best in the literature to date. We believe that reproduction of the experimental data at this level does provide a useful prediction of the highly excited states.

The results of the "ab initio" calculation [15] of the vibratio 1al constants and levels (column 9, Table 2) seem to be in qualitat ve agreement with our data. We do note that there are some large differences between the experimental values and the ab initio

BYKOV ET AL. 995

TABLE 4. The energy differences in cm⁻¹ between A1 and B1 stretching pairs

	n = 1	$E_{\rm B1}-E_{\rm A1}$	n = 2	$E_{\rm B1}-E_{\rm A1}$	n = 3	$E_{\rm B1}-E_{\rm A1}$	n = 4	$E_{\rm B1}-E_{\rm A1}$
$\overline{v} = 0$	(100, 001)	14.1	(200, 101)	2.3	(102, 201)	0.2	(202, 301)	0.0
v = 1	(110, 011)	10.1	(210, 111)	1.1	(112, 211)	0.0	(212, 311)	0.0
v = 2	(120, 021)	6.4	(220, 121)	0.3	(122, 221)	-0.0	(222, 321)	0.0
v = 3	(130, 031)	3.0	(230, 131)	-0.3	(132, 231)	-0.0	(232, 331)	0.0
v = 4	(140, 041)	-0.1	(240, 141)	-0.5	(142, 241)	-0.0	(242, 341)	0.0
v = 5	(150, 051)	-2.9	(250, 151)	-0.6	(152, 251)	0.0	(252, 351)	0.0

^{*}v is the number of bending quanta of v_2 , n is the total number of stretching quanta of v_1 and v_3 . The stretching pairs are shown in parentheses.

calculations for the (040) state and other higher vibrational states involving the bending vibration. The levels involving low excitation of the bending mode ($\nu_2 < 3$) calculated by Kozin and Jensen [3] using the variational MORBID (Morse Oscillator Rigid Bender Internal Dynamics) method (column 8, Table 2) agree within 1 cm⁻¹ with our results. For states involving the excited bending vibration, the differences between our and the Kozin, Jensen calculations range from 2 cm⁻¹ for the (040) state to 30 cm⁻¹ for the highest bending state (080). Note that in the case of our calculations, the agreement for the (040) state is satisfactory because its observed upper state level was used in our fitting, while in the Kozin, Jensen paper the (040) energy value is purely a prediction.

The MORBID approach uses the exact vibration-rotation Hamiltonian with the intramolecular potential energy function having a reasonable asymptotic behavior. Energy levels are calculated by a "direct" numerical diagonalization, and the potential energy function parameters are fitted to a large number of rotational-vibrational levels of four isotopic species of hydrogen sulfide. The MORBID calculations tend to give accurate energy levels; hence the good agreement between our levels and those predicted within the MORBID approach is evidence of the validity of the effective vibrational Hamiltonian method.

5. The local modes in hydrogen sulfide

5.1. Vibrational energy spectrum of H₂S and local mode limit

The traditional theory of vibrational-rotational spectra is based on the concept of normal coordinates and a perturbation treatment of vibration-rotation interactions with anharmonic corrections and relevant effective rotational Hamiltonians [16]. The conventional approach has been successful in explaining the spectra caused by transitions to low-lying vibrational states, but other approaches are required for highly excited vibrational states. The local mode model has been successfully applied to fit the vibrational spectra of H₂X, XH₃, XH₄, and several other types of molecules (see, for instance, refs. 17-28). In this treatment, the molecule is represented as the sum of independent Morse oscillators with a weak potential and kinetic couplings between them while the bending vibration is frozen. The local mode model does explain the spacing between states of symmetric and asymmetric vibrational mode (the local mode pair) and the degeneracy of these levels under high excitation.

An extensive set of H₂S vibrational-rotation energies (experimental and calculated) gives us a unique opportunity to

understand the local mode limit in some detail. We first note that the relation predicted by Lehmann [26] and described by Mills and Robiette [27] is satisfied by the fitted H₂S parameters shown in Table 3.

$$x_{11} = x_{33} = 0.25x_{13} = \Gamma_{DD}$$

or

$$-24.3 \approx -24.5 \approx -23.7 \approx -23.3$$

Secondly, we can determine the energy where the "local mode limit" is reached so that the stretching modes become degenerate. The energy differences between stretching pairs $(E_{\rm B1}-E_{\rm A1})$ are presented in Table 4. These clearly demonstrate the influence of the bending vibration on the stretching pair spacing. When bending vibration is frozen (column 1 in the Table 4) the stretching pairs become degenerate beginning at n=3. When the number of the stretching quanta are small, the increase of the bending vibration quanta leads to a sharp decrease of $E_{\rm B1}-E_{\rm A1}$. When the number of the stretching quanta exceeds two, the bending vibration does not affect the degeneracy of the pairs.

5.2. Rotational energy levels at the local mode limit

Despite extensive studies of local mode vibrational motion in different types of molecules and the theoretical analysis of rotational structure of an isolated mode doublet, there is only limited information about the rotation—vibration energy structure for the case of local mode behavior (see ref. 25). Existing models employ simplifying assumptions about the intramolecular potential energy function [23]. Little has been reported about the rotational structure in the local mode limit.

In the present study, we examined the rotation-vibration energy levels of the nearly degenerate pairs of vibrational states corresponding to the excitation of three and four quanta of the stretching vibration. Band origins and rotational constants were determined by fitting experimental energy levels. For example, the calculated and observed upper state levels, the differences between observed and calculated values, and the mixing coefficients for the {(301)-(202)} and {(311)-(212)} local mode pairs are presented in Tables 5 and 6, respectively. We noted some interesting features. First, all levels of the stretching pairs are at least doubly degenerate, despite the fact that they are determined from different sets of lines of parallel and perpendicular bands. In the first pair, the degenerate pairs consist of one level belonging to (301) and another one belonging to

TABLE 5. Upper state energy levels (in cm⁻¹) and mixing coefficients for the (301) and (202) vibrational states of H₂³²S

Fig.	(301)							(202)			
1	$J K_a K_c$	$E_{ m calcd}$	$E_{ m obsd}$	O-C	Mi	xing	$E_{ m caled}$	$E_{ m obsd}$	O-C	N	lixing
1 1 9 925,371 9 925,369 -1 81,30 17,0 9 925,371 17,0 83,0 10,0					100.0	0.0					100.0
1 0 9 9292250 9 9292253 2 1000 0.0 9 929250 9 900 1000 1			9 923.856				9 923.854	9 923.849	-4		
2 0 2 9 947,148 9 947,143 - 5 544 456 9 947,148 9 947,141 - 7 45.5 54.5 2 1 1 9 958,779 9 958,782 2 8.29 17.1 9 958,779 9 958,772 - 6 17.1 82.9 2 1 1 9 958,779 9 958,782 2 8.29 17.1 9 958,779 9 958,772 - 6 17.1 82.9 3 958,744 9 958,772 - 6 17.1 82.9 17.1 9 958,779 9 958,772 - 6 17.1 9 958,779 9 958,772 - 6 17.1 9 958,779 9 958,772 - 6 17.1 9 958,779 9 958,772 - 6 17.1 9 958,779 9 958,772 - 6 17.1 9 958,779 9 958,772 - 6 17.1 9 958,779 9 958,772 - 6 17.1 9 958,779 9 958,772 - 6 17.1 9 958,779 9 958,772 - 6 17.1 9 958,779 9 958,772 - 6 17.1 9 958,779 9 958,772 - 6 17.1 9 958,779 9 958,772 - 6 17.1 9 958,779 9 958,772 - 6 17.1 9 958,779 9 9 9 8 17.1 9 958,779 9 9 9 8 17.1 9 958,779 9 9 9 8 17.1 9 958,779 9 9 9 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1						17.0					
2 1 2 946.784 946.777 -6 53.4 46.6 946.784 946.784 -2 46.7 53.3 2 1 10 995.792 995.792 -6 17.1 82.9 946.784 946.784 946.784 -7 17.1 82.9 946.784 946.784 946.784 -7 17.1 82.9 946.784 946.784 -7 17.1 82.9 946.784 946.784 -7 17.1 82.9 946.784 -7 17.1 82.9 946.784 -7 17.1 82.9 946.784 -7 17.1 82.9 946.784 -7 17.1 82.9 946.784 -7 17.1 82.9 946.784 -7 17.1 82.9 946.784 -7 17.1 82.9 946.784 -7 17.1 82.9 -7 17.1 82.9 -7 17.1 82.9 -7 17.1 82.9 -7 17.1 82.9 -7 17.1 82.9 -7 17.1 82.9 -7 17.1 82.9 -7 17.1 82.9 -7 17.1 82.9 -7 17.1 82.9 -7 17.1 82.9 -7 17.1 82.9 -7 17.1 82.9 -7 17.1 82.9 -7 17.1 82.9 -7 17.1 17.1 82.9 -7 17.1				_ 			9 929.230 0 047 148	9 947 141	_7		
2 1 1 9 958.779 9 958.782 2 8.29 17.1 9 958.779 9 958.772 -6 17.1 82.9 2 2 0 9 956.0323 9 956.323 9 956.323 9 958.315 -7 17.1 82.9 3 0 3 9 9578.21 9 956.323 9 956.323 9 958.315 -7 17.1 82.9 3 0 3 9 9578.21 9 956.323 9 958.323 9 958.315 -7 17.1 82.9 3 1 2 10000.776 10000.772 -3 548 143 9 958.22 3 1 1 0000.881 10000.772 -3 548 143 9 958.22 3 2 1 10000.881 10000.772 -3 548 145.2 3 3 0 1002.387 10000.381 5 82.2 17.8 10000.881 10000.897 -4 95.6 4.4 1002.965 10000.387 100000.387 10000.387 10000.387 10000.387 10000.387 10000.387 10000.387 10000.387 10000.387 10000.387 10000.387 10000.387 10000.387 10000.387 10000.387 10000.387 10000.387 10000.387 10000.387 1000000.387 10000000000.387 1000000000000000000000000000000000000	$\frac{2}{2} \frac{0}{1} \frac{2}{2}$										53.3
2 2 1 9 963.323 9 996.322 0 82.9 17.1 9 963.323 9 963.315 -7 17.1 82.9 89.9 3 0 3 9 978.421 0 986.046 0 98.9 1.1 9 966.046 1 1.1 98.9 9 6.047 1 1 1 82.9 9 6.047 1 1 1 82.9 9 6.047 1 1 1 1 82.9 9 6.047 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	2 1 1		9 958.782				9 958,779	9 958,772			
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3 1 2 10001,776 10001,777 -3 54.8 45.2 10001,776 10001,776 -4 71.8 54.9 3 2 2 10000,881 10000,779 -1 50.4 49.6 10000,881 10000,887 5 46.5 50.4 3 3 3 1 1000,887 1001,887 -1 45.1 54.9 54.1 1000,887 -1 45.1 54.9 54.1 1000,887 -1 45.1 54.9 54.1 1000,887 -1 45.1 54.9 54.1 1000,887 -1 60.0 6	2 2 0					1.1					98.9
3 1 2 10 001.776 10 001.772 -3 54.8 45.2 10 001.776 10 001.774 -1 45.1 54.9 3 2 1 10 001.375 10 011.381 5 82.2 17.8 10 011.375 10 011.375 -4 17.8 82.2 3 3 0 10 020.387 10 020.386 0 82.5 17.5 10 020.387 10 011.370 -4 17.8 82.2 3 3 0 10 020.387 10 020.386 0 82.5 17.5 10 020.387 10 011.370 -4 17.8 82.2 4 1 4 1 00.3765 10 021.960 -4 95.6 4.4 10 021.966 10 021.985 -6 4.4 95.6 4 1 4 1 00.8779 10 018.739 -2 99.4 0.7 0.0 0.0 0.0 0.0 4 1 3 10 050.739 10 018.739 -2 0.0 0.0 0.0 0.0 0.0 4 1 3 10 050.739 10 050.731 0 0.	3 0 3	9 978.421				13.4					
3 2 2 10 000.081 10 000.079	3 1 3					15.2		10.001.774			
3 2 1 10011.375 10011.381 5 82.2 17.8 10011.375 10011.375	3 1 2									45.1 40.6	
3 3 1 10020,387 10020,386 0 82.5 17.5 10020,387 10012,955 -6 4.4 95.6 4.4 0 10018,757 10018,758 0 99.4 0.6 10018,757 10018,752 -4 0.6 99.4 4 1 4 10018,757 10018,758 0 99.4 0.6 10018,757 10018,752 -4 0.6 99.4 4 1 4 10018,757 10018,758 0 0.7 10018,759 10018,752 -4 0.6 99.4 1.4 1 10018,757 10018,759											82.2
3 3 0 10021,965 10021,960 -4 95.6 4.4 10021,968 10021,958 -6 4.4 9 5.6 40 4 10018,737 10018,732 -4 0.6 99.4 4 1 4 10018,749 10018,732 2 99.3 0.7 10018,739 10018,752 -4 0.6 99.3 4 1 4 1 3 10050,634 10050,634 0 85.7 14.3 10050,634 10050,616 -16 14.3 85.7 44.2 3 10050,635 10050,634 0 85.7 14.3 10050,635 10050,616 -16 14.3 85.7 44.2 2 3 10050,252 10050,251 10050,251 10050,251 10050,252 10050,252 10050,251 10050,251 10074,782 10050,252 10050,252 -6 15.8 84.2 42.2 2 10074,300 10070,290 -1 45.9 54.1 10070,300 10070,290 -1 45.9 54.1 10070,300 10070,300 -0 54.1 45.9 45.9 44.4 1 3 1008,965 1008,965 2 39.8 20.2 1008,966 10070,300 10070,300 10070,300 -1 55.5 44.5 10050,966 10050,						17.5		10 0111270	·		82.4
4 1 4 10 018,749 10 018,752 2 99,3 0.7 10 018,758 8 0.7 99,3 4 1 3 10 050,634 10 050,634 0 85,7 14,3 10 050,633 10 050,635 10 50,0252 10 050,251 10 074,782 10 074,780 -1 55,5 44,5 10 074,782 10 050,245 -6 15,8 84,2 4 2 2 10 074,780 10 074,780 -1 55,5 44,5 10 074,782 10 050,245 -6 15,8 84,2 4 3 1 10 074,780 10 074,780 -1 55,5 44,5 10 074,782 44,5 55,5 44,5 10 074,782 10 074,780 10 07									-6	4.4	95.6
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6 0 6 10 126.252								10 102 512	_1		
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6 3 4 10 216.306						4.4		10 176.187	14		
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6 4 3 10 244.849 71.5 28.5 10 244.850 10 244.833 16 28.5 71.5 6 4 2 10 261.842 10 261.837 -4 60.1 39.9 10 261.843 39.9 60.1 6 5 2 10 276.432 41.3 39.9 10 276.432 10 276.432 0 58.7 41.3 6 5 1 10 279.171 10 279.176 4 65.7 34.3 10 279.171 34.3 41.3 6 6 1 10 306.924 10 306.926 1 77.0 23.0 10 307.062 10 307.077 15 19.8 80.2 7 0 7 10 193.376 10 193.377 0 98.9 1.1 10 193.376 10 193.375 0 51.0 49.0 10 193.377 10 193.377 0 49.0 10 193.377 10 193.377 0 1.1 98.9 7 1 6 10 252.363 -3 94.8 5.2						4.2					
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0.00.000	8 2 7										
	8 2 6	10 395.818			86.6						

TABLE 5. (concluded)

		(301)					(202)			
$J K_a K_c$	Ecalcd	$E_{ m obsd}$	O-C	Mi	xing	$E_{\rm calcd}$	$E_{ m obsd}$	O-C	М	ixing
8 3 6	10 395.808	10 395.794	-13	86.8	13.2	10 395.808	10 395.808	0	13.4	86.6
8 3 5	10 444.553			91.9	8.1	10 444.552			8.2	91.8
8 4 5	10 444.312			91.4	8.6	10 444.313			8.6	91.4
8 4 4	10 484.219	10 484.200	-18	90.1	9.9	10 484.217			9.9	90.1
8 5 4	10 481.399			82.6	17.4	10 481.399			17.4	82.6
8 5 3	10 504.923	10 504.929	5	46.0	54.0	10 504.924			54.0	46.0
8 6 3	10 553.512			53.9	46.1	10 553.512			46.1	53.9
8 6 2	10 552.867			64.5	35.5	10 552.867			35.5	64.5
8 7 2	10 524.359			44.1	55.9	10 524.360			55.9	44.1
8 7 1	10 518.150	10 518.160	9	78.7	21.3	10 518.149			21.3	78.7
8 8 1	10 593.728			71.2	28.8	10 593.728			28.8	71.2
8 8 0	10 593.746	10 593.745	0	72.2	27.8	10 593.747			27.8	72.2
9 0 9	10 354.317	10 354.316	0	55.3	44.7	10 354.317	10 354.321	3	44.3	55.7
9 1 9	10 354.317	10 354.316	0	55.7	44.3	10 354.317	10 354.321	3	44.7	55.3
9 1 8	10 431.377	10 431.385	7	98.9	1.1	10 431.377	10 431.385	7	46.3	53.7
9 2 8	10 431.377	10 431.372	-4	53.7	46.3	10 431.375	10 431.372	-2	1.1	98.9
9 2 7	10 498.712			81.1	18.9	10 498.711			28.4	71.6
9 3 7	10 498.708	10 498.708	0	71.6	28.4	10 498.709			18.9	81.1
9 3 6	10 556.341			83.2	16.8	10 556.341			17.4	82.6
9 4 6	10 556.290			82.6	17.4	10 556.291			16.9	83.1
9 4 5	10 604.340			88.4	11.6	10 604.341			11.6	88.4
9 5 5	10 603.557			87.0	13.0	10 603.558			13.0	87.0
9 5 4	10 644.004			85.3	14.7	10 644.005			14.7	85.3
9 6 4	10 637.862			71.8	28.2	10 637.864			28.2	71.8
9 6 3	10 659.950			51.8	48.2	10 659.949			48.2	51.8
9 7 3	10 719.412			61.4	38.6	10 719.412			38.6	61.4
9 7 2	10 719.135			67.7	32.3	10 719.134			32.3	67.7
9 8 2	10 683.614			49.9	50.1	10 683.615			50.1	49.9
9 8 1	10 680.044			76.4	23.6	10 680.045			23.6	76.4
9 9 1	10 765.521			68.2	31.8	10 765.521			31.8	68.2
9 9 0	10 765.526			68.7	31.3	10 765.527			31.3	68.7

^{*}O-C = observed - calculated band centers in units of 10^{-3} cm⁻¹.

(202). The degeneracy is within $0.018\ cm^{-1}$ for levels listed in Table 5.

To model the observed levels, the effective rotational Hamiltonian is written in the usual manner; apart from the Watson-type "diagonal" Hamiltonian, it contains the Coriolis-type "resonance" operator. If we use the Hamiltonian with different rotational and centrifugal distortion constants for the two vibrational states belonging to the given local mode pair, we find that the parameters of the two states are very similar. For instance, the rotational constants of the (301) and (202) pair are $9.615 98 \pm 0.000 60, 8.614 83 \pm 0.000 71, 4.476 45 \pm$ 0.000 13 and 9.615 89 ± 0.000 69, 8.613 15 ± 0.000 77, 4.47677 ± 0.00017 for the A, B, and C constants, respectively, and the centrifugal distortion constants are equally close. If we force the rotational constants of the pairs to be equal, reducing the number of fitted parameters from 17 to 9, the quality of the fitting does not change very much; in the first case, the standard deviation is 0.005 cm⁻¹ and for the second it is 0.006 cm⁻¹. Thus, one can conclude that the highly excited local mode vibrations lead to an alignment of rotational and centrifugal distortion constants for the paired states and that the parameters become identical. As a consequence one can use for the calculations the simple models with fewer adjusted parameters. The fitted parameters for these states are presented in Table 7. The Hamiltonian reproduces the experimental data well enough to assign unambiguously the spectrum recorded at 0.02 cm⁻¹ resolution. For 120 energy levels of (301)–(202) and 129 levels of (311)–(212), the standard deviations are 0.006 cm⁻¹, and the largest deviation is 0.017 cm⁻¹.

As seen from Tables 5 and 6, the mixing between rotational sublevels is strong. There are numerous levels with approximately fifty-fifty mixing, and the maximum of the mixing shifts toward the larger values of the K_a quantum number with increasing J. This causes the strengthening of the weak component of local mode pair (e.g., (202)). As a consequence, there are a large number of lines that are actually doublets. This leads to some difficulties in the energy levels determination. We also found examples of fourfold clustering in the (301)–(202) pair starting with J = 4 levels at $K_a = 0$ and 1 of both states. For J = 6, the degeneracy is 0.001 cm⁻¹, and analogous clustering takes place for $K_a = 1$ and 2 levels starting at J = 6, and for $K_a = 2$ and 3 at J = 8 etc. This kind of fourfold clustering was previously predicted by Lehmann [23] and later confirmed by Kozin and Jensen during MORBID calculations for the ν_1/ν_3 bands of H₂S [3] and H₂Se [28]. This type of clustering is formed by "coexistence" of two energy doublets belonging to the states of a local mode pair. The present result is experimental evidence of this phenomena at low J and K_{α} values.

6. Conclusion

The large set of rotational–vibration energy levels obtained from the high-resolution Fourier transform spectra of H_2S up to 11 147 cm⁻¹ has permitted detailed studies of its vibrational structure in the local mode limit. The high excited bending vibrational states (030), (040), (050), (130), and (031) have been analyzed for the first time so that the bending–vibrational interactions could be investigated. The local mode limit is clearly

TABLE 6. Upper state energy levels (in cm⁻¹) and mixing coefficient for the (311) and (212) vibrational states of H₂³²S

		(311)					(212)			=======================================
$J K_a K_c$	$E_{ m calcd}$	$E_{ m obsd}$	O-C	Mi	xing	$E_{ m calcd}$	$E_{ m obsd}$	O-C	М	ixing
0 0 0	11 008.684	11 008.684	0	100.0	0.0	11 008.684			0.0	100.0
1 0 1	11 021.667	11 021.669	l	83.3	16.7	11 021.667	11 021.670	2	16.7	83.3
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	11 023.299 11 027.451	11 023.301 11 027.451	1	83.3 100.0	16.7 0.0	11 023.299 11 027.451	11 027.459	7	16.7 0.0	83.3 100.0
$\frac{1}{2} \frac{1}{0} \frac{0}{2}$	11 045.107	11 027.431	ő	54.0	46.0	11 045.107	11 027.439	7	46.1	53.9
2 1 2	11 044.714	11 044.711	$-\tilde{2}$	52.8	47.2	11 044.713	11 044.716	2	47.1	52.9
2 1 1	11 057.555	11 057.555	0	83.3	16.7	11 057.555			16.7	83.3
$\begin{array}{cccc} 2 & 2 & 1 \\ 2 & 2 & 0 \end{array}$	11 062.443 11 065.352	11 062.441 11 065.355	$-1 \\ 3$	83.3 98.9	16.7 1.1	11 062.443 11 065.352	11 062.440	-2	16.7 1.1	83.3 98.9
3 0 3	11 076.331	11 076.336	4	86.3	13.7	11 076.331	11 076.335	3	13.9	86.1
3 1 3	11 076.270	11 076.272	$\dot{2}$	85.9	14.1	11 076.270	11 076.270	ő	13.9	86.1
3 1 2	11 099.519	11 099.518	0	50.1	49.9	11 099.519	11 099.517	-1	49.9	50.1
3 2 2	11 101.350	11 101.349	0	45.7	54.3	11 101.351	11 101.356	5	54.3	45.7
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	11 111.607 11 121.299	11 111.604 11 121.299	$-3 \\ 0$	82.5 82.8	17.5 17.2	11 111.607 11 121.299	11 111.604 11 121.299	$-3 \\ 0$	17.5 17.2	82.5 82.8
3 3 0	11 122.979	11 122.980	0	95.6	4.4	11 122.979	11 122.979	ő	4.4	95.6
4 0 4	11 116.508	11 116.510	ĺ	99.5	0.5	11 116.508	11 116.506	-1	0.5	99.5
4 1 4	11 116.500	11 116.506	5	99.5	0.5	11 116.500	11 116.510	9	0.5	99.5
4 1 3 4 2 3	11 150.649	11 150.651	1	85.8	14.2	11 150.649	11 150.641	$^{-8}$	14.2	85.8
4 2 3 4 2 2	11 150.235 11 176.514	11 150.232 11 176.511	$-2 \\ -2$	84.3 55.0	15.7 45.0	11 150.234 11 176.514	11 150.237	2	15.7 45.0	84.3 55.0
4 3 2	11 171.673	11 171.665	-7	45.5	54.5	11 171.674	11 171.672	1	54.5	45.5
4 3 1	11 184.167	11 184.165	1	80.1	19.9	11 184.167	11 184.166	0	19.9	80.1
4 4 1	11 199.956	11 199.957	0	81.6	18.4	11 199.956	11 199.955	0	18.4	81.6
4 4 0 5 0 5	11 200.779	11 200.778	0	90.5 96.7	9.5 3.3	11 200.779	11 200.783 11 165.523	3 5	9.5 0.9	90.5 99.1
5 0 5 5 1 5	11 165.518 11 165.516	11 165.524 11 165.524	6 8	90.7 99.1	3.3 0.9	11 165.518 11 165.517	11 165.524		3.3	99.1 96.7
5 1 4	11 209.280	11 209.286	5	98.7	1.3	11 209.281	11 105.521	,	1.3	98.7
5 2 4	11 209.211	11 209.210	0	98.5	1.5	11 209.211			1.5	98.5
5 2 3 5 3 3	11 243.293	11 243.297	3	84.6	15.4	11 243.294	11 243.293	0	15.4	84.6
5 3 3 5 3 2	11 241.761 11 270.725	11 241.756 11 270.732	$-4 \\ 7$	79.5 56.3	20.5 43.7	11 241.761 11 270.726	11 241.753 11 270.721	−7 −4	20.5 43.7	79.5 56.3
5 4 2	11 261.189	11 261.181	-8	41.7	58.3	11 261.190	11 2/0./21	-4	58.3	41.7
5 4 1	11 275.769	11 275.775	6	74.7	25.3	11 275.767	11 275.765	- 1	25.3	74.7
5 5 1	11 298.449	11 298.449	0	79.8	20.2	11 298.449	11 298.449	0	20.2	79.8
5 5 0	11 298.809	11 298.811	1	85.2	14.8	11 298.809	11 298.809	0	14.8	85.2
6 0 6 6 1 6	11 223.335 11 223.336	11 223.333 11 223.333	$-1 \\ -2$	99.0 99.3	1.0 0.7	11 223.335 11 223.336	11 223.333 11 223.333	-1 -2	0.7 1.0	99.3 99.0
6 1 5	11 276.820	11 276.817	$-\frac{2}{2}$	91.7	8.3	11 276.821	11 276.819	$-\frac{2}{1}$	4.9	95.1
6 2 5	11 276.810	11 276.819	9	95.1	4.9	11 276.811	11 276.817	6	8.3	91.7
6 2 4	11 320.090	11 320.100	9	97.0	3.0	11 320.091			3.0	97.0
6 3 4	11 319.755	11 319.744	10	96.0	4.0	11 319.755	11 319.744	-10	4.0	96.0
6 3 3 6 4 3	11 354.265 11 350.244	11 354.269 11 350.243	3	83.0 71.6	17.0 28.4	11 354.265 11 350.243	11 354.262 11 350.247	$-2 \\ 3$	17.0 28.4	83.0 71.6
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	11 384.109	11 384.114	4	58.3	41.7	11 384.108	11 330.247	3	41.7	58.3
6 5 2	11 368.395	11 368.402	7	39.7	60.3	11 368.396	11 368.403	7	60.3	39.7
6 5 1	11 387.005	11 387.009	3	65.9	34.1	11 387.004	11 416 720	0	34.1	65.9
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	11 416.738 11 416.885	11 416.881	-3	77.4 80.5	22.6	11 416.738 11 416.884	11 416.738	0	22.6 19.5	77.4 80.5
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	11 289.949	11 289.949	0	54.4	19.5 45.6	11 289.948	11 289.949	0	0.5	99.5
7 1 7	11 289.949	11 289.949	ŏ	99.5	0.5	11 289.949	11 289.949	õ	45.6	54.4
7 1 6	11 353.158	11 353.152	-5	61.7	38.3	11 353.157	11 353.152	-4	15.5	84.5
7 2 6	11 353.156	11 353.152	-3	84.5	15.5	11 353.156	11 353.152	-3	38.3	61.7
7 2 5 7 3 5	11 406.014 11 405.951	11 405.958	6	92.3 92.6	7.7 7.4	11 406.014 11 405.952			7.3 7.8	92.7 92.2
7 3 4	11 448.766	11 405.750	Ü	94.2	5.8	11 448.765			5.8	94.2
7 4 4	11 447.621	11 447.610	-10	91.1	8.9	11 447.621			8.9	91.1
7 4 3	11 483.626			81.2	18.8	11 483.626			18.8	81.2
7 5 3	11 475.351			62.4	37.6	11 475.349			37.6	62.4
7 5 2 7 6 2	11 516.747 11 493.798			60.9 40.4	39.1 59.6	11 516.748 11 493.798			39.1 59.6	60.9 40.4
7 6 1	11 518.215			55.8	44.2	11 518.213			44.3	55.7
7 7 1	11 554.719	11 554.720	0	74.6	25.4	11 554.718			25.4	74.6
7 7 0	11 554.772	11 245 250		76.3	23.7	11 554.773	11 554.786	12	23.7	76.3
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	11 365.350 11 365.349	11 365.350 11 365.350	0	99.7 99.7	0.3 0.3	11 365.351 11 365.349	11 365.350 11 365.350	0	0.3 0.3	99.7 99.7
8 1 7	11 438.269	11 438.263	-5	99.7 98.8	1.2	11 438.268	11 438.263	-4	45.2	54.8
8 2 7	11 438.268	11 438.263	-4	54.8	45.2	11 438.268	11 438.263	$-\dot{4}$	1.2	98.8
8 2 6	11 500.747			81.1	18.9	11 500.747			15.2	84.8

TABLE 6. (concluded)

		(311)					(212)			
$J K_a K_c$	$E_{\rm calcd}$	$E_{ m obsd}$	O-C	Mix	king	$E_{ m calcd}$	$E_{ m obsd}$	O-C	M	ixing
8 3 6 8 3 5 8 4 5 8 4 4 8 5 4 8 5 3 8 6 3 8 6 2 8 7 2	11 500.735 11 552.837 11 552.584 11 595.198 11 592.157 11 617.204 11 669.334 11 668.659 11 638.025 11 631.477	11 500.728	-7	84.8 90.3 89.9 90.4 82.9 45.7 53.5 64.0 44.1 79.1	15.2 9.7 10.1 9.6 17.1 54.3 46.5 36.0 55.9 20.9	11 500.736 11 552.838 11 552.584 11 595.197 11 592.157 11 617.203 11 669.333 11 668.660 11 638.026 11 631.478	11 500.728	-8	18.9 9.7 10.1 9.6 17.1 54.3 46.5 36.0 55.9 20.9	81.1 90.3 89.9 90.4 82.9 45.7 53.5 64.0 44.1 79.1
8 8 1 8 8 0	11 712.238 11 712.259			71.7 72.6	28.3 27.4	11 712.238 11 712.259			28.3 27.4	71.7 72.6

TABLE 7. Fitted constants (in cm⁻¹) of local mode pairs of H₂S

	(301) and (202)	(311) and (212)
$E_{\rm v}$	9911.022 5	11008.683 6
A	9.614 81 (29)	9.929 16 (51)
В	8.615 33 (33)	8.841 91 (19)
C	4.476 61 (12)	4.415 53 (61)
$\Delta_k \times 10^3$	3.990 (16)	4.067 (34)
$\Delta a \times 10^{\circ}$	-2.563(10)	$-2.368 \stackrel{?}{2} (44)$
$\Delta_{\rm j}^{\rm j} \times 10_{\rm A}^{\rm 4}$	6.531 (12)	6.560 9 (51)
$\overline{\delta_k}^{\rm J} \times 10^4$	-3.155(67)	-6.0
$\delta_i \times 10^4$	2.950 (10)	2.970 5 (29)
$\delta_{\rm j} \times 10^4$ $H_{\rm k} \times 10^6$	1.8	2.83 (53)
C_{xz}	0.569 651 (88)	-0.608 171 (87)
Standard deviation	$0.006 \mathrm{cm}^{-1}$	0.005
Number of levels	120	129

Note: Estimated uncertaintes are $1\,\sigma$ in the last digit. Parameters without uncertainties were held fixed.

demonstrated in several highly excited vibrational states: (211)-(112), (301)-(202), and (311)-(212). The rotational structure of the stretching pairs is completely degenerate from J=0 up to J=10, and the spacing between corresponding levels in the stretching pairs is less than $0.02 \, \mathrm{cm}^{-1}$. This allows the mixing of the spectroscopic parameters of one vibrational component with another component of the stretching pair. As a result, a fourfold degeneracy of the $K_a=0$ and $K_a=1$ arises at low J values when strong excitation of the local modes occurs. Refinement of the rotational constants and analyses of dipole moment parameters are currently in progress and will be reported later. Database predictions based on ref. 8 and the present results are available for the 4 μ m region.

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